

Periodic Table

Observe the p-block elements of the periodic table and answer the questions that follow.

The periodic table is divided into four blocks based on the orbitals occupied by the valence electrons. The valence electrons of p-block elements occupy the p-orbital

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5						
13	Al					
31						
49						
	Indium 114.82	Tin 118.71	Antimony 121.76	Tellurium 127.60	Iodine 126.90	Xenon 131.29
81	Tl Thallium 204.38	Pb Lead 207.21	Bi Bismuth 208.98	Po Polonium (209)	At Astatine (210)	Rn Radon (222)
113	Nh Nihonium (286)	Fl Flerovium (289)	Mc Moscovium (289)	Lv Livermorium (293)	Ts Tennessine (294)	Og Oganesson (294)

Periodic Table

1) Write the chemical symbols of the elements below.

a) Antimony –

b) Tin –

c) Bromine –

d) Sulfur –

2)

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3)

d) None of the above

4) Label each element as a metal, nonmetal, or metalloid.

a) Aluminum –

a) Germanium –

c) Carbon –

c) Gallium –

Periodic Table

5) Write the atomic mass and atomic number of phosphorus.

a) Atomic mass _____

b) Atomic number _____

6)

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7)

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